

## 2-Methyl-5-[3-(2-methylphenyl)-1,2,4-oxadiazol-5-ylmethylsulfanyl]-1,3,4-thiadiazole

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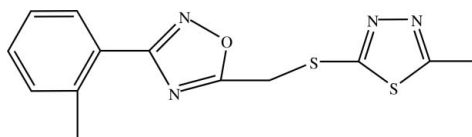
Received 28 October 2007; accepted 30 October 2007

Key indicators: single-crystal X-ray study;  $T = 293$  K; mean  $\sigma(\text{C}-\text{C}) = 0.007$  Å;  $R$  factor = 0.062;  $wR$  factor = 0.178; data-to-parameter ratio = 15.1.

In the title compound,  $\text{C}_{13}\text{H}_{12}\text{N}_4\text{OS}_2$ , the central oxadiazole ring makes dihedral angles of 8.9 (2) and 89.6 (3)° with the benzene and thiadiazole rings, respectively.

### Related literature

For related literature, see: Nicolaides *et al.* (1998); Romero (2001); Talar & Dejai (1996).



### Experimental

#### Crystal data

$\text{C}_{13}\text{H}_{12}\text{N}_4\text{OS}_2$   
 $M_r = 304.39$   
 Monoclinic,  $P2_1/c$   
 $a = 10.309$  (2) Å  
 $b = 12.992$  (3) Å  
 $c = 10.600$  (2) Å  
 $\beta = 100.63$  (3)°

$V = 1395.3$  (5) Å<sup>3</sup>  
 $Z = 4$   
 Mo  $K\alpha$  radiation  
 $\mu = 0.38$  mm<sup>-1</sup>  
 $T = 293$  (2) K  
 0.30 × 0.10 × 0.10 mm

#### Data collection

Enraf–Nonius CAD-4  
 diffractometer  
 Absorption correction:  $\psi$  scan  
 (North *et al.*, 1968)  
 $T_{\min} = 0.894$ ,  $T_{\max} = 0.963$   
 2890 measured reflections

2737 independent reflections  
 1694 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.032$   
 3 standard reflections  
 every 200 reflections  
 intensity decay: none

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.062$   
 $wR(F^2) = 0.178$   
 $S = 1.03$   
 2737 reflections

181 parameters  
 H-atom parameters constrained  
 $\Delta\rho_{\max} = 0.29$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.35$  e Å<sup>-3</sup>

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Siemens, 1996); software used to prepare material for publication: *SHELXL97*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB2618).

### References

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**supplementary materials**

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## 2-Methyl-5-[3-(2-methylphenyl)-1,2,4-oxadiazol-5-ylmethylsulfanyl]-1,3,4-thiadiazole

P.-L. Wang, H.-L. Li, S.-S. Kang, H.-S. Zeng and H. Wang

### Comment

1,2,4-Oxadiazoles represent an important class of five-membered heterocycles with medicinal applications such as anti-inflammatory (Nicolaidis *et al.*, 1998) and antipicornaviral (Romero, 2001) properties. We are focusing our synthetic and structural studies on new oxindole derivatives. Sulfurether derivatives exhibited strong inhibiting activity to *Staphylococcus aureus* (Talar & Dejai, 1996). We report here the structure of the title compound, (I), containing a thiadiazole sulfanylether group.

There are three rings in the molecule. The benzene and oxadiazole ring are close to coplanar due to the extended aromatic system [dihedral angle = 8.9 (2)°]. The angle between the oxadiazole plane and the thiadiazole moiety is 89.6 (3)°. The molecular structure of (I) is shown in Fig. 1.

### Experimental

5-Mercapto-2-methyl-1,3,4-thiadiazole (20 mmol) was dissolved in ethanol (70 ml) and water (70 mmol). Sodium acetate (20 mmol) was added to this mixture. Then 3-[2-(methyl)phenyl]-5-chloromethyl-1,2,4-oxadiazol (40 mmol) was added. The resulting mixture was refluxed for 8 h. After cooling and filtrating, the crude title compound was obtained. Pure compound (I) was obtained by crystallizing from a mixture of ethyl acetate (8 ml) and petrolum ether (bp. 333–363 K) (4 ml). Colourless blocks of (I) were obtained by slow evaporation of an ethanol solution. <sup>1</sup>H NMR (CDCl<sub>3</sub>, δ, p.p.m.): 7.35–7.36 (m, 1H), 7.10–7.13(m, 3H), 4.18–4.19 (s, 2H), 2.38–2.39 (s,3H), 2.32–2.33 (s,3H).

### Refinement

The H atoms were placed geometrically (C—H = 0.93–0.96 Å) and refined as riding with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  or  $1.5U_{\text{eq}}(\text{methyl C})$ .

### Figures

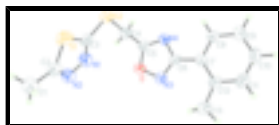


Fig. 1. A view of the molecular structure of (I), showing displacement ellipsoids at the 30% probability level (H atoms represented by arbitrary spheres).

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### Crystal data

C<sub>13</sub>H<sub>12</sub>N<sub>4</sub>OS<sub>2</sub>

$F_{000} = 632$

# supplementary materials

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$M_r = 304.39$

Monoclinic,  $P2_1/c$

Hall symbol: -P 2ybc

$a = 10.309$  (2) Å

$b = 12.992$  (3) Å

$c = 10.600$  (2) Å

$\beta = 100.63$  (3)°

$V = 1395.3$  (5) Å<sup>3</sup>

$Z = 4$

$D_x = 1.449$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation

$\lambda = 0.71073$  Å

Cell parameters from 25 reflections

$\theta = 9$ – $13^\circ$

$\mu = 0.38$  mm<sup>-1</sup>

$T = 293$  (2) K

Block, colourless

$0.30 \times 0.10 \times 0.10$  mm

## Data collection

Nonius CAD-4  
diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

$T = 293$ (2) K

$\omega/2\theta$  scans

Absorption correction:  $\psi$  scan  
(North *et al.*, 1968)

$T_{\min} = 0.894$ ,  $T_{\max} = 0.963$

2890 measured reflections

2737 independent reflections

1694 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.032$

$\theta_{\text{max}} = 26.0^\circ$

$\theta_{\text{min}} = 2.0^\circ$

$h = -12 \rightarrow 12$

$k = 0 \rightarrow 16$

$l = 0 \rightarrow 13$

3 standard reflections

every 200 reflections

intensity decay: none

## Refinement

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.062$

$wR(F^2) = 0.178$

$S = 1.03$

2737 reflections

181 parameters

Primary atom site location: structure-invariant direct  
methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring  
sites

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.060P)^2 + 2.3P]$$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} < 0.001$

$\Delta\rho_{\text{max}} = 0.29$  e Å<sup>-3</sup>

$\Delta\rho_{\text{min}} = -0.35$  e Å<sup>-3</sup>

Extinction correction: none

## Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
O	0.1434 (4)	0.1828 (2)	0.0705 (3)	0.0684 (10)
S1	0.53469 (12)	0.38060 (11)	0.12681 (13)	0.0669 (4)
N1	0.2889 (4)	0.3910 (3)	0.0430 (4)	0.0695 (12)
C1	0.4913 (6)	0.5355 (4)	0.3034 (5)	0.0754 (15)
H1B	0.4209	0.5770	0.3239	0.113*
H1C	0.5582	0.5793	0.2804	0.113*
H1D	0.5286	0.4947	0.3766	0.113*
S2	0.39670 (13)	0.24762 (10)	-0.09151 (13)	0.0618 (4)
N2	0.3157 (4)	0.4621 (3)	0.1412 (4)	0.0640 (11)
C2	0.4386 (4)	0.4665 (3)	0.1935 (4)	0.0492 (10)
N3	0.1246 (5)	0.0994 (3)	0.1516 (4)	0.0687 (12)
C3	0.3944 (4)	0.3437 (3)	0.0250 (4)	0.0506 (11)
N4	0.2023 (3)	0.0434 (3)	-0.0210 (3)	0.0462 (8)
C4	0.2240 (4)	0.2126 (4)	-0.1264 (4)	0.0526 (11)
H4B	0.1706	0.2743	-0.1294	0.063*
H4C	0.2041	0.1801	-0.2101	0.063*
C5	0.1892 (4)	0.1415 (3)	-0.0289 (4)	0.0436 (9)
C6	0.1615 (4)	0.0195 (3)	0.0926 (4)	0.0423 (9)
C7	0.1625 (4)	-0.0868 (3)	0.1392 (4)	0.0421 (9)
C8	0.2228 (5)	-0.1601 (3)	0.0731 (4)	0.0520 (11)
H8A	0.2607	-0.1397	0.0041	0.062*
C9	0.2273 (5)	-0.2620 (4)	0.1082 (5)	0.0612 (13)
H9A	0.2672	-0.3102	0.0628	0.073*
C10	0.1719 (5)	-0.2925 (4)	0.2116 (5)	0.0622 (13)
H10A	0.1757	-0.3611	0.2367	0.075*
C11	0.1115 (5)	-0.2215 (4)	0.2769 (5)	0.0592 (12)
H11A	0.0727	-0.2438	0.3446	0.071*
C12	0.1061 (4)	-0.1176 (3)	0.2458 (4)	0.0447 (10)
C13	0.0405 (5)	-0.0452 (4)	0.3241 (5)	0.0636 (13)
H13A	0.0105	-0.0827	0.3913	0.095*
H13B	-0.0335	-0.0131	0.2701	0.095*
H13C	0.1023	0.0067	0.3610	0.095*

*Atomic displacement parameters ( $\text{\AA}^2$ )*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
O	0.110 (3)	0.0366 (16)	0.074 (2)	0.0038 (17)	0.058 (2)	-0.0004 (15)
S1	0.0509 (7)	0.0809 (10)	0.0723 (8)	0.0051 (6)	0.0203 (6)	-0.0044 (7)
N1	0.062 (3)	0.055 (3)	0.097 (3)	0.003 (2)	0.028 (2)	-0.012 (2)

## supplementary materials

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C1	0.086 (4)	0.072 (4)	0.071 (3)	-0.008 (3)	0.021 (3)	-0.014 (3)
S2	0.0675 (8)	0.0563 (7)	0.0718 (8)	-0.0028 (6)	0.0393 (6)	-0.0030 (6)
N2	0.062 (3)	0.054 (2)	0.079 (3)	0.006 (2)	0.021 (2)	-0.010 (2)
C2	0.046 (2)	0.055 (3)	0.048 (2)	0.000 (2)	0.0133 (19)	0.008 (2)
N3	0.110 (3)	0.040 (2)	0.072 (3)	-0.001 (2)	0.059 (3)	0.000 (2)
C3	0.049 (2)	0.049 (3)	0.060 (3)	0.003 (2)	0.025 (2)	0.014 (2)
N4	0.064 (2)	0.0383 (19)	0.0408 (19)	0.0034 (17)	0.0222 (16)	-0.0005 (15)
C4	0.067 (3)	0.049 (3)	0.044 (2)	0.002 (2)	0.018 (2)	0.003 (2)
C5	0.046 (2)	0.048 (2)	0.040 (2)	0.0010 (19)	0.0138 (17)	-0.0027 (19)
C6	0.049 (2)	0.039 (2)	0.043 (2)	0.0028 (18)	0.0196 (18)	0.0031 (18)
C7	0.047 (2)	0.039 (2)	0.043 (2)	-0.0015 (18)	0.0151 (18)	0.0009 (18)
C8	0.068 (3)	0.042 (2)	0.054 (3)	-0.003 (2)	0.031 (2)	-0.003 (2)
C9	0.077 (3)	0.042 (3)	0.070 (3)	0.008 (2)	0.029 (3)	0.002 (2)
C10	0.070 (3)	0.037 (2)	0.083 (4)	-0.001 (2)	0.021 (3)	0.011 (2)
C11	0.056 (3)	0.061 (3)	0.066 (3)	-0.005 (2)	0.024 (2)	0.017 (2)
C12	0.041 (2)	0.050 (2)	0.045 (2)	0.0001 (19)	0.0127 (17)	0.0076 (19)
C13	0.073 (3)	0.064 (3)	0.064 (3)	-0.003 (3)	0.039 (3)	0.003 (2)

### Geometric parameters ( $\text{\AA}$ , $^\circ$ )

O—C5	1.343 (5)	C4—H4B	0.9700
O—N3	1.418 (5)	C4—H4C	0.9700
S1—C3	1.706 (5)	C6—C7	1.467 (5)
S1—C2	1.727 (5)	C7—C8	1.394 (6)
N1—C3	1.293 (5)	C7—C12	1.420 (5)
N1—N2	1.381 (6)	C8—C9	1.373 (6)
C1—C2	1.491 (6)	C8—H8A	0.9300
C1—H1B	0.9600	C9—C10	1.383 (6)
C1—H1C	0.9600	C9—H9A	0.9300
C1—H1D	0.9600	C10—C11	1.370 (7)
S2—C3	1.759 (5)	C10—H10A	0.9300
S2—C4	1.809 (5)	C11—C12	1.389 (6)
N2—C2	1.287 (6)	C11—H11A	0.9300
N3—C6	1.304 (5)	C12—C13	1.496 (6)
N4—C5	1.283 (5)	C13—H13A	0.9600
N4—C6	1.382 (5)	C13—H13B	0.9600
C4—C5	1.479 (6)	C13—H13C	0.9600
C5—O—N3	106.1 (3)	N3—C6—N4	113.6 (4)
C3—S1—C2	87.5 (2)	N3—C6—C7	125.0 (4)
C3—N1—N2	111.8 (4)	N4—C6—C7	121.4 (3)
C2—C1—H1B	109.5	C8—C7—C12	119.6 (4)
C2—C1—H1C	109.5	C8—C7—C6	116.8 (4)
H1B—C1—H1C	109.5	C12—C7—C6	123.7 (4)
C2—C1—H1D	109.5	C9—C8—C7	121.2 (4)
H1B—C1—H1D	109.5	C9—C8—H8A	119.4
H1C—C1—H1D	109.5	C7—C8—H8A	119.4
C3—S2—C4	100.5 (2)	C8—C9—C10	119.5 (4)
C2—N2—N1	113.5 (4)	C8—C9—H9A	120.3
N2—C2—C1	123.4 (4)	C10—C9—H9A	120.3

N2—C2—S1	112.8 (3)	C11—C10—C9	120.0 (4)
C1—C2—S1	123.7 (4)	C11—C10—H10A	120.0
C6—N3—O	103.5 (3)	C9—C10—H10A	120.0
N1—C3—S1	114.3 (4)	C10—C11—C12	122.5 (4)
N1—C3—S2	124.1 (4)	C10—C11—H11A	118.8
S1—C3—S2	121.6 (3)	C12—C11—H11A	118.8
C5—N4—C6	103.6 (3)	C11—C12—C7	117.2 (4)
C5—C4—S2	111.8 (3)	C11—C12—C13	118.8 (4)
C5—C4—H4B	109.3	C7—C12—C13	123.9 (4)
S2—C4—H4B	109.3	C12—C13—H13A	109.5
C5—C4—H4C	109.3	C12—C13—H13B	109.5
S2—C4—H4C	109.3	H13A—C13—H13B	109.5
H4B—C4—H4C	107.9	C12—C13—H13C	109.5
N4—C5—O	113.2 (4)	H13A—C13—H13C	109.5
N4—C5—C4	129.0 (4)	H13B—C13—H13C	109.5
O—C5—C4	117.6 (4)		
C3—N1—N2—C2	0.1 (6)	O—N3—C6—N4	-0.4 (5)
N1—N2—C2—C1	-178.5 (4)	O—N3—C6—C7	179.1 (4)
N1—N2—C2—S1	-0.8 (5)	C5—N4—C6—N3	0.2 (5)
C3—S1—C2—N2	1.0 (4)	C5—N4—C6—C7	-179.3 (4)
C3—S1—C2—C1	178.6 (4)	N3—C6—C7—C8	-171.0 (5)
C5—O—N3—C6	0.4 (5)	N4—C6—C7—C8	8.4 (6)
N2—N1—C3—S1	0.6 (5)	N3—C6—C7—C12	9.0 (7)
N2—N1—C3—S2	179.6 (3)	N4—C6—C7—C12	-171.6 (4)
C2—S1—C3—N1	-0.9 (4)	C12—C7—C8—C9	0.9 (7)
C2—S1—C3—S2	-179.9 (3)	C6—C7—C8—C9	-179.1 (4)
C4—S2—C3—N1	-17.8 (4)	C7—C8—C9—C10	-0.5 (8)
C4—S2—C3—S1	161.1 (3)	C8—C9—C10—C11	1.0 (8)
C3—S2—C4—C5	-80.5 (3)	C9—C10—C11—C12	-1.8 (8)
C6—N4—C5—O	0.1 (5)	C10—C11—C12—C7	2.1 (7)
C6—N4—C5—C4	176.4 (4)	C10—C11—C12—C13	-178.8 (5)
N3—O—C5—N4	-0.4 (5)	C8—C7—C12—C11	-1.6 (6)
N3—O—C5—C4	-177.1 (4)	C6—C7—C12—C11	178.4 (4)
S2—C4—C5—N4	-83.2 (5)	C8—C7—C12—C13	179.3 (4)
S2—C4—C5—O	92.9 (4)	C6—C7—C12—C13	-0.7 (7)

Fig. 1

